

St. Lawrence University **Physics 308**  
**Syllabus, Spring 2019**  
 Daniel W. Koon, Instructor  
 Last updated: 1/11/19

**TEXT:**

[Introduction to Electrodynamics](#), (4th Ed.) David J. Griffiths. ([Ebay](#), [Amazon](#)) A copy is on reserve in the Science Library.

**OTHER FINE TEXTS:**

[Electricity and Magnetism](#), Edward M. Purcell & David J. Morin.  
[Electromagnetic Fields and Waves](#), Lorrain, Corson, and Lorrain.

**“YSBATs” (You Should Be Able To...):** Exam preparation material [To be posted on Sakai site]

**HOMEWORK:** There will be two types of homeworks: a shorter problem that is usually quite directly associated with the material in a particular lecture -- called the “Daily” -- and a longer set of more thoughtful problems that will generally require more out-of-class time to work through -- the regular, weekly “Homework” assignment. Both sets of homework assignments will be posted on Sakai, and later, the solutions will also be posted. Homework or Dailies submitted after solutions have already been posted will earn less than full credit, if any.

**ATTENDANCE POLICY:** The instructor reserves the right to subtract one-half of a letter grade for each absence beyond the first three. I do not expect you to learn all of your physics from my magnificent lectures, just as I would not expect you to learn it all simply from reading every sentence of the text, or by doing every homework problem. However, each one of these items you miss diminishes the understanding you are likely to gain of the course material.

**EXAMS:** There will be three exams during regular class time during the semester, plus a Final Exam. (See Calendar below for dates and times.)

**GRADING POLICY:** Your final average will be calculated from the following:

Weekly Homework	40%
‘Dailies’ (daily homework)	12%
3x semester Exams	36% (12% each)
Final Exam	12%

Grades will be posted in the “PostEm” folder at Sakai, with the final numerical grade determined by the following scale:

4.0	94%+	3.0	82%+	2.0	70%+	1.0	60%+
3.75	91%+	2.75	79%+	1.75	67%+		
3.50	88%+	2.50	76%+	1.50	64%+		
3.25	85%+	2.25	73%+	1.25	61%+		

The following Excel code will calculate your numerical grade, “y” from your percentage, “x”:  
 $y = IF(x < 60, 0, MIN(4, INT((x - 46)/3)/4))$

All cutoffs are exact.

**ACADEMIC RESOURCES, SPECIAL NEEDS:** If you have a disability and need accommodations please be sure to contact the Disability and Accessibility Services Office (x5537) right away so they can help you get the accommodations you require. If you will need to use any accommodations in this class, please talk with me early so you can have the best possible experience this semester. Although not required, I would like to know of any accommodations that are needed at least 10 days before a quiz or test, so please see me soon. For more specific information visit the DASO website <http://www.stlawu.edu/disability-and-accessibility-services>.

## THE CALENDAR

**Physics 308 – Electricity and Magnetism, St. Lawrence University**  
**Dr. Daniel W. Koon • Last revised: 1/11/19.**

**THE LECTURES:** (Chapters refer to Griffiths' *Introduction to Electrodynamics*, unless otherwise stated. The pre-lecture notes are available on Sakai.)

Date	Topic	Chapter	Friday HW assignment
1-17 R	1. Intro, Coulomb, Superposition	"Advertisement" ( <i>pp. xiv-xviii</i> ), Ch. 2.1	
1-22 T	2. E-field, integrating continuous charge distributions	Ch. 2.1.3-4	
1-24 R	3. Field lines & Flux, Gauss' Law, applications	Ch. 2.2.1-3	HW#1
1-29 T	4. Del, Gauss's Theorem, Stokes' Theorem & work	Ch. 1.2, 2.2.1-4	
1-31 R	5. Electric Potential, Poisson's Equation	Ch. 2.3.1-3	HW#2
2-5 T	6. Summary, Work and energy in electrostatics	Ch. 2.3-4	
2-7 R	7. Conductors and capacitors	Ch. 2.5	HW#3
<b>2-12 T</b>	<b>EXAM I</b>	<b>Lecture 1-6</b>	
<b>2-14 R</b>	<b>MID-SEMESTER BREAK</b>		
2-19 T	8. Energy of a capacitor, Poisson's Equation: Relaxation	Ch. 2.5.4, 3.1.1-5	
2-21 R	9. Poisson's Equation: Relaxation & Images Art	This week's notes (Sakai), Ch. 3.2: <a href="#">1.</a> <a href="#">2.</a> <a href="#">Relaxation</a>	HW#4
2-26 T	10. Poisson: Separation of variables, Multipole expansion	Ch. 3.3.2, 3.4.1, <a href="#">Images handout</a>	
2-28 R	11. Multipoles, esp. dipole fields	Ch. 3.4.2-4	HW#5
3-5 T	12. Dielectrics	This week's notes, plus Skim Ch. 4	
3-7 R	13. Lorentz force, Current, Right hand rules	Ch. 5.1	HW#6
<b>3-12 T</b>	<b>EXAM II</b>	<b>Lecture 7-12</b>	
3-14 R	14. Right hand rules, Magnetism, Biot & Savart	Ch. 5.2	HW#7
<b>3-18 to 3-22</b>	<b>SPRING BREAK</b>		
3-26 T	15. Ampère's Law, current density, monopoles	Ch. 7.1.1-2: Ch. 5.3, 5.1.3:	
3-28 R	16. Emf, Ohm's law, drift velocity		HW#8
4-2 T	17. Rowland, resistivity, Hall effect	Handouts, this week's notes, Purcell Ch. 6 handout	
4-4 R	18. Motional Emf, Faraday's Law	Ch. 7.1.3-7.2.2	HW#9
4-9 T	19. Lenz' Law, Inductance, Magnetic energy	Ch. 7.3: Ch. 7.2.3-4	
4-11 R	20. Maxwell's Equations, symmetry	Continue	HW#10
<b>4-16 T</b>	<b>EXAM III</b>	<b>Lecture 13-19</b>	
4-18 R	21. Solutions to Maxwell's Equations, electromagnetic waves	9.2	
4-23 T	22. Waves in matter, Snell's law, Fresnel's Eqs.	9.3, <a href="#">Fresnel handout</a>	
4-25 R	23. Dipole radiation, Rayleigh scattering	Ch. 11.1.2 & correct? → <a href="#">Rayleigh handout</a>	HW#11
4-30 T	24. E-fields of moving particles	Ch. 12.3.2, Purcell, Ch. 5 (Sakai)	
5-2 R	25. Electrostatics + relativity = magnetostatics	Ch. 12.3.1, Purcell, Ch. 5 (Sakai)	HW#12
<b>FINAL EXAM</b>	<b>Wednesday, May 8 at 1:30pm</b>		

## ACADEMIC HONESTY: SELECTIONS FROM THE SLU STUDENT HANDBOOK

All students at St. Lawrence University are bound by honor to maintain the highest level of academic integrity. By virtue of membership in the St. Lawrence community, every student accepts the responsibility to know the rules of academic honesty, to abide by them at all times, and to encourage all others to do the same.

Responsibility for avoiding behavior or situations from which academic dishonesty may be inferred rests entirely with the students. Claims of ignorance, unintentional error, and academic or personal pressure are not excuses for academic dishonesty. Students should be sure to learn from faculty what is expected as their own work and how the work of other people should be acknowledged. Instructors are expected to maintain conditions which promote academic honesty.

Instructors have the duty to investigate any instance involving possible academic dishonesty and must present evidence of academic dishonesty to the Academic Honor Council rather than make private arrangements with the student involved. Violations of the St. Lawrence University Code of Academic Honor are administered under the constitution of the Academic Honor Council [See Student Handbook for the Constitution].

### Academic Honesty

The primary objective of the University is the promotion of knowledge. This objective can be furthered only if there is strict adherence to scrupulous standards of honesty. At St. Lawrence, all members of the University community have a responsibility to see that standards of honesty and integrity are maintained. Students who respect academic honesty and who are orderly and meticulous in their treatment of both their own work and the work of others should anticipate no difficulty with cheating, plagiarism, or other forms of academic dishonesty. Borrowing ideas or language from others is acceptable scholarly practice and in many instances actively to be encouraged.

Academic dishonesty generally arises from one of two sources: either a student has knowingly cheated or plagiarized or he/she has been careless or slipshod in discriminating between his/her own work and that of others or in acknowledging sources accurately. These latter difficulties are easily circumvented. Any standard handbook on English usage or term paper writing manual will furnish a methodology as well as appropriate internal reference, endnote, or bibliographical forms (cf., for example, the *Harbrace Handbook*, *A Guide to MLA Documentation*, or *Writers Inc.*).

A major objective of the University is the pursuit of knowledge which can be achieved only by strict adherence to standards of honesty. At St. Lawrence, all members of the community have a responsibility to see that these standards are maintained.

### Academic Dishonesty

1. It is assumed that all work submitted for credit is done by the student unless the instructor gives specific permission for collaboration.
2. Cheating on examinations and tests consists of knowingly giving or using or attempting to use unauthorized assistance during examinations or tests.
3. Dishonesty in work outside of examinations and tests consists of handing in for credit as original work that which is not original, where originality is required.

### The following constitute examples of academic dishonesty:

- a. Plagiarism: Presenting as one's own work the work of another person - words, ideas, data, evidence, thoughts, information, organizing principles, or style of presentation-without proper attribution. Plagiarism includes paraphrasing or summarizing without acknowledgment by quotation marks, footnotes, endnotes, or other indices of reference (cf. Joseph F. Trimmer, *A Guide to MLA Documentation*).
- b. Handing in false reports on any experiment.
- c. Handing in a book report on a book one has not read.
- d. Falsification of attendance records of a laboratory or other class meeting.
- e. Supplying information to another student knowing that such information will be used in a dishonest way.
- f. Submission of work (papers, journal abstracts, etc.) which has received credit in a previous course to satisfy the requirement(s) of a second course without the knowledge and permission of the instructor of the second course.

Claims of ignorance and academic or personal pressure are unacceptable as excuses for academic dishonesty. Students must learn what constitutes one's own work and how the work of others must be acknowledged.

St. Lawrence students are required to sign the following statement prior to registration for classes:

*"I hereby acknowledge that I have read the above document and I understand my responsibility in maintaining the standards of academic honesty at St. Lawrence University."*